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(54) Title: METHOD OF MAKING PHOTOCATALYSTS BY LOADING TITANIUM DIOXIDE FILM ON FLEXIBLE SUB-
STRATES

(57) Abstract: Disclosed are methods of making a photocatalyst by loading titanium dioxide film on a flexible substrate, comprising the steps of: (1) preparing an active layer sol-gel by: (a) making a precursor solution comprising n-butyl titanate, ethanol, diethanolamine, and water; (b) adding a pore-forming agent selected from the group consisting of polyglycol, octadecylamine, and mixtures thereof to the precursor solution; and (c) placing the resulting solution in a sealed gelatinization process for at least 3 days; and (2) preparing an active TiO₂ photocatalyst layer by: (a) coating a flexible substrate with the active layer sol-gel prepared according to step (1) using a pulling and coating process; (b) drying the coated flexible substrate; and (c) placing the coated, dried flexible substrate in a hydrothermal kettle for thermal crystallization in a mixed solvent of ethanol and water at 60-200°C. Further disclosed are methods wherein the precursor solution comprises titanium tetrachloride, ethanol, and water.

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METHOD OF MAKING PHOTOCATALYSTS BY LOADING TITANIUM DIOXIDE FILM ON FLEXIBLE SUBSTRATES

FIELD OF THE INVENTION

The present invention relates to a method of making photocatalysts, especially a method of making photocatalysts by loading titanium dioxide film on a flexible substrate, and the photocatalyst made thereby.

BACKGROUND OF THE INVENTION

At present, there are essentially three known methods for manufacturing surface-load titanium dioxide (TiO_2) photocatalysts: (1) using sol-gels to form a TiO_2 film directly on the substrate and undergoing high-temperature calcination; (2) dispersing nano-powder in a suspension solution, loading it onto the substrate, and undergoing high-temperature calcination; and (3) using inorganic or organic gels to load nano photocatalysts onto metal screens. The TiO_2 photocatalytic films manufactured by sol-gel process of the method (1) have no pores, small specific surface areas, and low activity. In addition, the calcination temperature is usually over 400°C , so the substrate must be resistant to high temperatures. The photocatalytic films manufactured according to the method (2) tend to peel off easily because the bonding between the secondary powder and the substrate is weak. Consequently, this method is of little practical value. The photocatalytic effectiveness of the catalyst manufactured according to the method (3) is reduced

because the catalytic films are wrapped up by inorganic or organic sol-gels. The bonding between the films and the substrates is weak. In addition, organic sol-gels are likely to have UV decomposition.

The aforementioned methods usually employ sheet materials (such as metal plates and glass plates) or glass beads as photocatalytic supports. The photocatalysts thus manufactured have some shortcomings, such as limited areas of effective light exposure, limited areas of contact between photocatalysts and fluids, and great air resistance unfavorable for high flow rate reaction. In addition, the substrate materials are likely to diffuse into the photocatalysts, thus reducing the activity of the photocatalysts and making it hard to form active crystalline phase structures. Photocatalysts currently available generally employ honeycomb ceramics as supports to overcome the disadvantages of sheet or pellet supports in applications. Ceramic supports, however, have disadvantages, too. First, they are expensive in cost and weak in mechanical strength, hence easy to break. Second, due to their rigidity, it is hard to manufacture ceramic photocatalytic components of specific structures or shapes. Third, the required manufacturing technology is so sophisticated that it is hard to produce large supports.

Chinese patent application numbers 01141902.4 and 01131093.6 disclose surface-load medium-size pore TiO_2 nano films on substrates of glass beads and metal screens by sol-gel processes of spinning off excessive sol-gel and high temperature calcination. The substrates disclosed in these references are readily available and low in cost. The

photocatalysts so manufactured are believed to have strong bonding strength, be easy to manufacture, versatile in application, and highly effective. However, as these manufacturing processes require a temperature of 350-550°C, they are not suitable for non-woven fabrics, woven fabrics, dust-free paper and other flexible substrate materials that are not resistant to high temperatures.

Thus there remains a need for low temperature methods by which photocatalytic substrates can be made from flexible substrate materials such as non-woven fabrics, woven fabrics, dust-free paper and other flexible substrate materials that are not resistant to high temperatures. The present invention provides such methods.

SUMMARY OF THE INVENTION

The present invention relates to methods of making a photocatalyst by loading titanium dioxide film on a flexible substrate, comprising the steps of: (1) Preparing an active layer sol-gel by: (a) Making a precursor solution comprising n-butyl titanate, ethanol, diethanolamine, and water; (b) Adding a pore-forming agent selected from the group consisting of polyglycol, octadecylamine, and mixtures thereof to the precursor solution; and (c) Placing the resulting solution in a sealed gelatinization process for at least 3 days; and (2) Preparing an active TiO₂ photocatalyst layer by: (a) Coating a flexible substrate with the active layer sol-gel prepared according to step (1) using a pulling and coating process; (b) Drying the coated flexible substrate; and (c) Placing the coated, dried flexible substrate in a hydrothermal kettle for

thermal crystallization in a mixed solvent of ethanol and water at 60-200°C. The present invention further relates to methods wherein the precursor solution comprises titanium tetrachloride, ethanol, and water.

DETAILED DESCRIPTION OF THE INVENTION

5 The present invention relates to methods of making flexible substrate surface-load titanium dioxide nanocrystalline film photocatalysts. Flexible material supports provide improved effectiveness of light utilization, increase the effective action areas
10 among the light, the photocatalyst and the fluids, and expand the applications of the photocatalysts. Flexible substrate materials are easy to obtain and low in cost. In addition, the methods according to the present invention utilize a thermo-solvent process to form active anatase structures at low temperatures. Therefore, non-woven fabrics,
15 woven fabrics, dust-free fabrics, and other flexible substrate materials that are not resistant to high temperatures can be used, providing reduced cost and expanding the practical applications of the photocatalytic substrates herein.

The present invention further relates to photocatalysts
20 manufactured according to the above methods.

The term "pulling and coating method", as used herein, means to pull the photocatalysts impregnated in sol-gels out of the sol-gels by using a pull apparatus. Excess portions of the sol-gels automatically fall back into the vessel containing the sol-gels under the action of gravity.
25 Portions of the sol-gels absorb on the surface of supports and form a

compact film layer. The thickness of the film is controlled via pulling speed, concentrate and viscosity of sol-gels so as to control the thickness of sol-gel film loaded on the supports and the thickness of photocatalyst layer formed.

5 The term "solvent thermal crystallization", as used herein, means that certain chemical products or materials are dissolved or dispersed in solvents (such as alcohol, water) and heat treated under a sealed conditions so that the temperature and pressure in a container are increased. When the pressure in the container is over 1 atmospheric
10 pressure, it can promote the chemical reactions or the formation of crystalline states that are difficult to be carried out under normal pressure, and achieve the object of forming crystalline phase under non-high temperature.

15 A preferred method of making flexible substrate surface-load titanium dioxide nanocrystalline film photocatalysts according to the present invention comprises the steps of: (1) Preparation of an active layer sol-gel; and (2) Preparation of an active photocatalyst layer. Each step is described in detail below.

(1) Preparation of an active layer sol-gel

20 A precursor solution is prepared as follows. Preferred precursors suitable for use in the present invention are n-butyl titanate and titanium tetrachloride, and mixtures thereof.

Using n-butyl titanate as a precursor, a precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1:8-
25 12:0.1-0.15:0.05-0.06 is prepared. The preferred addition sequence is:

water is added to ethanol solution, then diethanolamine as a stabilizing agent is added to the solution, n-butyl titanate solution is then added to the mixed solution to give a yellowish homogeneous clear solution, and then an organic additive as a pore-forming agent is added to the solution. Preferred pore-forming agents are polyglycol, octadecylamine, and mixtures thereof. The mass ratio of the amount of the pore-forming agent to the amount of the ethanol in the precursor solution is pore-forming agent : ethanol = 1% to 30%, preferably, 8% to 15%. The solution is placed in a sealed condition for at least 3 days, preferably from about 3 to about 7 days, to gelatinize, and a clear sol-gel is obtained.

Using titanium tetrachloride as a precursor, a precursor solution in the volume ratio of titanium tetrachloride : ethanol : water = 1:8-12:0.08-0.15 is prepared. The addition preferred sequence is: water is added to ethanol solution, then titanium tetrachloride is added to the solution to form a yellowish clear solution, and then an organic additive as a pore-forming agent is added to the solution. Preferred pore-forming agents are polyglycol, octadecylamine, and mixtures thereof. The mass ratio of the amount of the pore-forming agent to the amount of the ethanol in the precursor solution is pore-forming agent : ethanol = 1%-30%, preferably, 8-15%. The solution is placed in a sealed condition for at least 3 days, preferably from about 3 to about 7 days, and a clear sol-gel having a certain viscosity is obtained.

According to another preferred embodiment of the present invention, in the preparation of active layer sol-gel, an additional agent

selected from lanthanum nitrate, n-butyl silicate, and mixtures thereof, can be further added to the precursor solution at any time. The molar ratio of La to Ti is from 0% to about 5%, preferably from about 0.8% to about 1.2%; the molar ratio of Si to Ti is from 0% to about 40%, preferably from about 15% to about 25%. The action of lanthanum nitrate is believed to control the growth of TiO_2 nanocrystal so as to make the particle size of TiO_2 crystal at about 10-15 nm. The addition of n-butyl silicate is to form partial SiO_2 sol-gel in the TiO_2 sol-gel so as to control the growth of TiO_2 crystal and to increase the specific surface area of the photocatalysts.

(2) Preparation of an Active Photocatalyst Layer

The active layer sol-gel prepared according to step (1) is directly coated on a cleaned flexible substrate by pulling and coating method. Excess sol-gel is removed. The thickness of the sol-gel layer is controlled by adjusting the viscosity of the sol-gel and the number of pulling iterations. The resulting wet sol-gel film is dried and then placed in a hydrothermal kettle for thermal crystallization in a mixed solvent of ethanol and water preferably at a volume ratio of ethanol to water of 0-100% at 60-200°C, preferably for at least about 2 hours. To ensure the evenness and activity of TiO_2 film, the film is pulled one to four times, preferably 2-3 times.

According to another preferred embodiment of the present invention, in the preparation step of active photocatalyst layer, said excess sol-gel is removed by spinning or extrusion; said wet sol-gel film is dried preferably at 30-150°C, more preferably at 80-120°C.

According to still another preferred embodiment of the present invention, in the preparation step of active photocatalyst layer, the ratio (by volume) of ethanol to water in the mixed solvent of ethanol-water for solvent thermal crystallization is preferably from 0% to about 80%,
5 most preferably from 0% to about 20%; the temperature of solvent thermal crystallization is preferably from 120-140°C.

It should be noted that the temperature of solvent thermal crystallization has a great effect on the performance of the catalysts obtained. When the temperature is lower than 60°C, it is difficult to
10 form a perfect TiO_2 crystal structure and its activity is very low; contrarily, when the temperature is higher than 200°C, the flexible substrate may be sintered, carbonized or decomposed so that the structure of flexible substrate is destroyed. Therefore, it is necessary to select suitable solvent heat treatment temperature.

15 In the method of the present invention, the flexible substrate materials include non-woven fabrics, woven fabrics, dust-free paper, most preferably water-pricked non-woven fabrics which surfaces have strong hydrophilic property.

The flexible substrate TiO_2 nanocrystalline photocatalysts
20 manufactured according to the methods of the present invention have advantages of strong bonding strength, small gas resistance, high photocatalytic effectiveness and high activity. Throughout the entire preparation method, the raw materials used are low in cost, the processes are relatively simple, and the preparation temperatures are
25 low; therefore, the production cost is effectively reduced. It is believed

that the present invention has much practical value and application prospects.

BRIEF DESCRIPTION OF THE DRAWINGS

5 Fig. 1 is a SEM photograph of the combined state of the catalyst film of Example 1; and

Fig. 2 is a SEM photograph of the combined state of the catalyst film of Example 2.

EXAMPLES

10 The following examples further describe and demonstrate embodiments within the scope of the present invention. These examples are given solely for the purpose of illustration and are not to be construed as limitations of the present invention as many variations thereof are possible without departing from the spirit and scope.

15 In the following examples, the precursor (preferably titanium tetrachloride or n-butyl titanate), the pore-forming agent (preferably polyglycol or octadecylamine), solvent (preferably ethanol) and stabilizing agent (preferably diethanolamine) are commercially analytic pure or chemical pure products. The flexible substrate materials used
20 are non-woven fabrics, woven fabrics, and dust-free paper.

The photocatalytic performance of the catalysts obtained is evaluated via the following method: Photocatalytic reaction apparatus is comprised of a sleeve-type internal and external cylinder. A 8W
25 ultraviolet lamp at a wavelength of 254 nm is installed in the internal

sleeve. The internal sleeve is wrapped with a layer of flexible photocatalyst coating with TiO_2 photocatalyst, The average distance of the photocatalyst and the ultraviolet light source is 3 cm; its receiving light area is 112cm^2 . A certain concentrate of formaldehyde gas is entered from the internal slip and flowed out through a silk screen. The amount of formaldehyde in the outflow gas is determined by using gas chromatograph with a hydrogen flame detector.

Example 1:

(1) Preparation of the Active Layer Sol-gel: Using titanium tetrachloride as a precursor agent, prepare the precursor solution in the volume ratio of titanium tetrachloride : ethanol : water = 1 : 10 : 0.12. The addition sequence is as follows: first add water to the ethanol solution, then drip feed the titanium tetrachloride to produce a yellowish clear solution, and finally add 10% PEG400 (polyethylene glycol, molecular weight 400). Place the mixed solution in a sealed gelatinization process for 3 days and the resultant product is a yellowish clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of non-woven fabric with a cleaning agent, and then immerse the material in the active layer sol-gel prepared in step (1). After immersion for 1 minute, take the non-woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the non-woven fabric in the active layer sol-gel, take it out after 1 minute and spin off the sol-gel on its

surface, and then let it air-dry. Repeat this procedure until the non-woven fabric has had four active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO_2 sol-gel into a hydrothermal kettle with water as the solvent, heat the kettle to 110°C ,
5 keep at this temperature for 2 hours, take the fabric out, and then wash and oven-dry. The resultant product is a non-woven fiber substrate surface-load titanium dioxide film photocatalyst.

An electronic microscopic study (see Figure 1) has revealed that this photocatalytic film has strong bonding strength. A photocatalytic
10 property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 900ppm to 610ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

15

Example 2:

(1) Preparation of the Active Layer Sol-gel: Using titanium tetrachloride as a precursor agent, prepare the precursor solution in the volume ratio of titanium tetrachloride : ethanol : water = 1 : 12 : 0.15.

20 The addition sequence is as follows: first add water to the ethanol solution, then drip feed the titanium tetrachloride to produce a yellowish clear solution, and finally add 15% PEG400. Place the mixed solution in a sealed gelatinization process for 5 days and the resultant product is a yellowish clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of non-woven fabric with a cleaning agent, and then immerse the material in the active layer sol-gel prepared according to step (1). After immersion for 2 minutes, take the non-woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the non-woven fabric in the active layer sol-gel, take it out after 2 minutes and spin off the sol-gel on its surface, and then let it air-dry. Now the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO_2 sol-gel into a hydrothermal kettle with a mixed solvent of water and ethanol (volume ratio 1:1), heat the kettle to 140°C , keep at this temperature for 4 hours, take the fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

An electronic microscopic study (see Figure 2) has revealed that the photocatalytic film has strong bonding strength. A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 900ppm to 360ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 3:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the solution in the volume ratio of n-butyl

titanate : ethanol : diethanolamine : water = 1 : 10 : 0.12 : 0.06. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 20% PEG400 as a pore-forming agent to the solution. Place the mixed solution in a sealed gelatinization process for 7 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of non-woven fabric with a cleaning agent, and then immerse the material in the aforementioned active layer sol-gel. After immersion for 1 minute, take the non-woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the non-woven fabric in the active layer sol-gel, take it out after 1 minute and spin off the sol-gel on its surface, and then let it air-dry. Repeat this procedure until the non-woven fabric has had three active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO_2 sol-gel into a hydrothermal kettle with ethanol as the solvent, heat the kettle to 130°C , keep at this temperature for 2 hours, take the fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 900ppm to 450ppm at the

reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 4:

5 (1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed
10 the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 8% PEG400 as a pore-forming agent to the solution. Place the mixed solution in a sealed gelatinization process for 7 days and the resultant product is a clear sol-gel of a certain viscosity.

15 (2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of dust-free paper with a cleaning agent, and then immerse the material in the aforementioned active layer sol-gel. After immersion for 1 minute, take the dust-free paper out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then
20 let it air-dry. Re-immerses the dust-free paper in the active layer sol-gel, take it out after 1 minute and spin off the sol-gel on its surface, and then let it air-dry. Now the dust-free paper has had two active layers loaded on its surface. Finally, place the dust-free paper coated with wet TiO_2 sol-gel into a hydrothermal kettle with ethanol as the solvent, heat the
25 kettle to 120°C, keep at this temperature for 4 hours, take the paper out,

and then wash and oven-dry. The resultant product is a dust-free paper substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 900ppm to 560ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 5:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% PEG400 as a pore-forming agent to the solution. Place the mixed solution in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of woven fabric with a cleaning agent, and then immerse the material in the aforementioned active layer sol-gel. After immersion for 1 minute, take the woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the woven fabric in the active layer sol-gel,

take it out after 1 minute and spin off the sol-gel on its surface, and then let it air-dry. Now the woven fabric has had two active layers loaded on its surface. Finally, place the woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle with ethanol as the solvent, heat the kettle to
5 140°C, keep at this temperature for 3 hours, take the fabric out, and then wash and oven-dry. The resultant product is a woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the
10 concentration of a formaldehyde gas from 900ppm to 380ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 6:

15 (1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed
20 the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% PEG800 as a pore-forming agent to the solution. Place the mixed solution in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At the room temperature, wash a piece of woven fabric with a cleaning agent, and then immerse the material in the aforementioned active layer sol-gel. After immersion for 1 minute, take the woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the woven fabric in the active layer sol-gel, take it out after 1 minute and spin off the sol-gel on its surface, and then let it air-dry. Now the woven fabric has had two active layers loaded on its surface. Finally, place the woven fabric coated with wet TiO_2 sol-gel into a hydrothermal kettle with ethanol as the solvent, heat the kettle to 140°C , keep at this temperature for 3 hours, take the fabric out, and then wash and oven-dry. The resultant product is a woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 1000ppm to 100ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 7:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed

the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% PEG800 as a pore-forming agent to the solution. Place the mixed solution in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash and dry a piece of water-pricked non-woven fabric, and then immerse the material in the aforementioned active layer sol-gel. After immersion for 1 minute, take the non-woven fabric out, remove the excess sol-gel from its surface by extrusion, and then oven-dry it with 60°C air flows. Repeat the procedure until the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle with water as the solvent, heat the kettle to 130°C, keep at this temperature for 2 hours, take the non-woven fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 2000ppm to 100ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 8:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol
5 solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, finally add 10% PEG800 as a pore-forming agent and lanthanum nitrate with the La/Ti molar ratio at 1%. Place the mixed solution in a sealed gelatinization
10 process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash and dry a piece of non-woven fabric, and then immerse the material in the aforementioned active layer sol-gel. After
15 immersion for 1 minute, take the non-woven fabric out, remove the excess sol-gel from its surface by extrusion, and then oven-dry it with 60°C air flows. Repeat the procedure until the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle
20 with water as the solvent, heat the kettle to 130°C, keep at this temperature for 2 hours, take the non-woven fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the
25 photocatalyst has high catalytic activity and is capable of reducing the

concentration of a formaldehyde gas from 3000ppm to 50ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

5 **Example 9:**

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol
10 solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% PEG800 as a pore-forming agent and n-butyl silicate with the Si/Ti mole ratio of 15% to the solution. Place the mixed solution in a sealed
15 gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash and dry a piece of non-woven fabric, and then immerse the material in the aforementioned active layer sol-gel. After
20 immersion for 1 minute, take the non-woven fabric out, remove the excess sol-gel from its surface by extrusion, and then oven-dry it with 60°C air flows. Repeat the procedure until the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle
25 with water as the solvent, heat the kettle to 130°C, keep at this

temperature for 2 hours, take the non-woven fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

5 A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 3500ppm to less than 50ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

10 **Example 10:**

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol
15 solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% PEG800 as a pore-forming agent, and lanthanum nitrate with the La/Ti molar ratio at 1% and n-butyl silicate with the Si/Ti mole ratio at 20%
20 to the solution. Place the mixed solution place in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash and dry a piece of non-woven fabric, and then
25 immerse the material in the aforementioned active layer sol-gel. After

immersion for 1 minute, take the non-woven fabric out, remove the excess sol-gel from its surface by extrusion, and then oven-dry it with 60°C air flows. Repeat the procedure until the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle with water as the solvent, heat the kettle to 126°C, keep the temperature for 2 hours, take the non-woven fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 500ppm to less than 250ppm within 2 hours in a static reactor which has a volume of 500ml and a catalyst area of 10 cm² and uses natural sunlight as the light source for the reaction.

Example 11:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% PEG800 as a pore-forming agent to the solution. Place the mixed

solution in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of non-woven fabric with a cleaning agent, and then immerse the material in the active layer sol-gel prepared according to step (1). After immersion for 1 minute, take the non-woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the non-woven fabric in the active layer sol-gel, take it out after 1 minute and spin off the sol-gel on its surface, and then let it dry at 90°C. Now the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle with a mixed solvent of 50% water and 50% ethanol (volume ratio), heat the kettle to 90°C, keep at this temperature for 3 hours, take the fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 1000ppm to 300ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Example 12:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10% octadecylamine as a pore-forming agent to the solution. Place the mixed solution place in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of non-woven fabric with a cleaning agent, and then immerse the material in the active layer sol-gel prepared according to step (1). After immersion for 1 minute, take the non-woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it air-dry. Re-immerses the non-woven fabric in the active layer sol-gel, take it out after 1 minute and spin off the sol-gel on its surface, and then let it dry at 90°C. Now the non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO₂ sol-gel into a hydrothermal kettle with a mixed solvent of 80% water and 20% ethanol (volume ratio), heat the kettle to 130°C, keep at this temperature for 3 hours, take the fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the concentration of a formaldehyde gas from 1000ppm to 500ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light
5 source mainly of the 254nm wavelength.

Example 13:

(1) Preparation of the Active Layer Sol-gel: Using n-butyl titanate as a precursor agent, prepare the precursor solution in the volume ratio
10 of n-butyl titanate : ethanol : diethanolamine : water = 1 : 8 : 0.10 : 0.05. The addition sequence is as follows: first add water to the ethanol solution, then add diethanolamine as a stabilizing agent, then drip feed the n-butyl titanate solution into the aforementioned mixed solution to produce a yellowish homogeneous clear solution, and finally add 10%
15 PEG800 as a pore-forming agent to the solution. Place the mixed solution in a sealed gelatinization process for 5 days and the resultant product is a clear sol-gel of a certain viscosity.

(2) Preparation of the Active Photocatalyst Layer: At room temperature, wash a piece of non-woven fabric with a cleaning agent,
20 and then immerse the material in the active layer sol-gel prepared according to step (1). After immersion for 1 minute, take the non-woven fabric out, use a high-speed centrifugal spinner to spin off the sol-gel on its surface, and then let it dry at 90°C. Re-immerses the non-woven fabric in the active layer sol-gel, take it out after 1 minute and
25 spin off the sol-gel on its surface, and then let it dry at 90°C. Now the

non-woven fabric has had two active layers loaded on its surface. Finally, place the non-woven fabric coated with wet TiO_2 sol-gel into a hydrothermal kettle with a mixed solvent of 90% water and 10% ethanol (volume ratio), heat the kettle to 130°C , keep at this
5 temperature for 3 hours, take the fabric out, and then wash and oven-dry. The resultant product is a non-woven fabric substrate surface-load titanium dioxide film photocatalyst.

A photocatalytic property evaluation study has shown that the photocatalyst has high catalytic activity and is capable of reducing the
10 concentration of a formaldehyde gas from 2000ppm to 300ppm at the reaction flow rate of 160ml/min with an 8W UV lamp as the light source mainly of the 254nm wavelength.

Thus it can be seen that the flexible substrate surface-load nanocrystalline TiO_2 film photocatalysts made according to the present
15 invention have strong bonding strength, versatility in application, and high photocatalytic effectiveness. In addition, since the materials used in the present methods are inexpensive and the methods themselves are free from undue complexity, the present invention is believed to effectively lower production costs and provide substrates that have
20 much practical value and application.

What is claimed is:

1. A method of making a photocatalyst by loading titanium dioxide film on a flexible substrate, comprising the steps of:

5 (1)Preparing an active layer sol-gel by:

a) Making a precursor solution comprising n-butyl titanate, ethanol, diethanolamine, and water in the volume ratio of n-butyl titanate : ethanol : diethanolamine : water = 1:8-12:0.1-0.15:0.05-0.06;

10 b) Adding a pore-forming agent selected from the group consisting of polyglycol, octadecylamine, and mixtures thereof to the precursor solution of step (1) a), wherein the mass ratio of the amount of the pore-forming agent to the amount of the ethanol in the precursor solution is pore-forming agent : ethanol = 1%-30% : 1; and

15 c) Placing the resulting solution in a sealed gelatinization process for at least 3 days; and

(2)Preparing an active TiO₂ photocatalyst layer by:

20 a) Coating a flexible substrate with the active layer sol-gel prepared according to step (1) using a pulling and coating method;

b) Drying the coated flexible substrate; and

25 c) Placing the coated, dried flexible substrate in a hydrothermal kettle for thermal crystallization in a mixed solvent of ethanol and water at 60-200°C.

2. A method of making a photocatalyst by loading titanium dioxide film on a flexible substrate, comprising the steps of:

(1)Preparing an active layer sol-gel by:

- 5 a) Making a precursor solution comprising titanium tetrachloride, ethanol, and water in the volume ratio of titanium tetrachloride : ethanol : water = 1:8-12:0.08-0.15;
- 10 b) Adding a pore-forming agent selected from the group consisting of polyglycol, octadecylamine, and mixtures thereof to the precursor solution of step a), wherein the mass ratio of the amount of the pore-forming agent to the amount of the ethanol in the precursor solution is pore-forming agent : ethanol = 1%-30% : 1; and
- 15 c) Placing the resulting solution in a sealed gelatinization process for at least 3 days; and

(2)Preparing an active TiO₂ photocatalyst layer by:

- 20 a) Coating a flexible substrate with the active layer sol-gel prepared according to step (1) using a pulling and coating method;
- b) Drying the coated flexible substrate; and
- c) Placing the coated, dried flexible substrate in a hydrothermal kettle for thermal crystallization in a mixed solvent of ethanol and water at 60-200°C.

25 3. The method according to claim 1 or 2, wherein in the step of

preparing said active layer sol-gel further comprises adding an additional agent selected from the group consisting of lanthanum nitrate, n-butyl silicate, and mixtures thereof, to the precursor solution.

5

4. The method according to claim 3, wherein the molar ratio of lanthanum to titanium is from 0% to about 5%.

10

5. The method according to claim 4, wherein the molar ratio of lanthanum to titanium is from about 0.8% to about 1.2%.

6. The method according to claim 3, wherein the molar ratio of silica to titanium is from 0% to about 40%.

15

7. The method according to claim 6, wherein the molar ratio of silica to titanium is from about 15% to about 25%.

8. The method according to any of claims 1-7 wherein the ratio of said pore-forming agent to ethanol is from about 8% to about 15%.

20

9. The method according to any of claims 1-7 wherein said flexible substrate is selected from the group consisting of non-woven fabrics, woven fabrics, dust-free papers, water-pricked non-woven fabrics having strong surface hydrophilic property, and mixtures thereof.

25

10. The method according to any of claims 1-7 further comprising the step of removing excess sol-gel by spinning or extrusion, after the step of coating the flexible substrate with the active layer sol-gel prepared according to step (1).

11. The method according to any of claims 1-7 wherein step of drying the coated flexible substrate is carried out at a temperature of from about 30°C to about 150°C.

12. The method according to claim 11 wherein said drying temperature is from about 80°C to about 120°C.

13. The method according to any of claims 1-7 wherein the ratio of ethanol to water in the mixed solvent of ethanol and water used in the thermal crystallization step is from 0% to about 80%.

14. The method according to claim 13 wherein the ratio of ethanol to water is from 0% to about 20%.

15. The method according to any of claims 1-7 wherein said thermal crystallization is carried out at a temperature of from about 120°C to about 140°C.

16. A photocatalyst prepared by the method according to any one of claims 1-15.

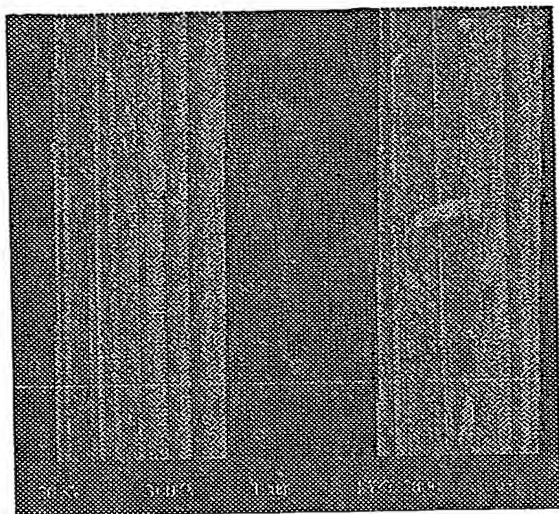


Figure 1

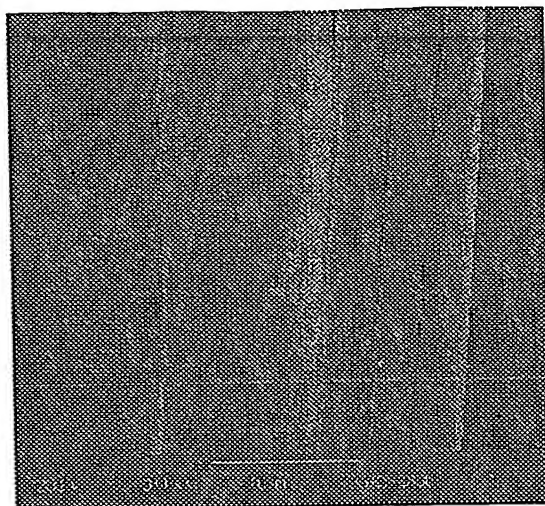


Figure 2

INTERNATIONAL SEARCH REPORT

International application No.
PCT/CN03/00553

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 B01J21/06

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 B01J C01B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

WPI,PAJ,EPODOC,CNPAT,中文期刊数据库

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP0826633 A1 (SHOWA DENKO KK et al) 04.Mar 1998 (04.03.98) Whole document	1-16
A	CN1342518 A (UNIV QINGHUA) 03.Apr 2002 (03.04.02) Whole document	1-16

☐ Further documents are listed in the continuation of Box C. ☒ See patent family annex.

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- "P" document published prior to the international filing date but later than the priority date claimed

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Date of the actual completion of the international search
30.Sep 2003

Date of mailing of the international search report

23 OCT 2003 (23.10.03)

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INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.
PCT/CN03/00553

EP0826633	A1	04.03.98	US6479031	B2	12.11.02
			JP11043327	A	16.02.99
			US2001016264	A1	23.08.01
			US6340711	B1	22.01.02
			EP0826633	B1	13.03.02
			US2002042449	A1	11.04.02
			DE69710975	E	18.04.02
CN 1342518	A	03.04.02	NONE		

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